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Key indicators

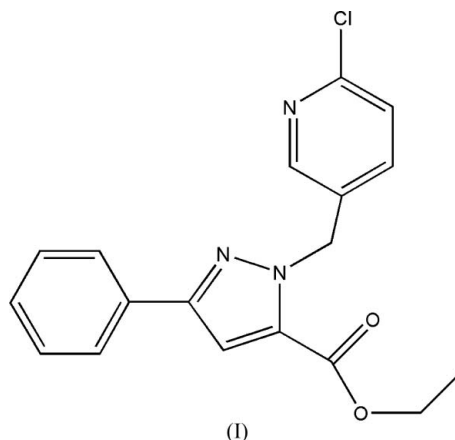
Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.045
 wR factor = 0.115
Data-to-parameter ratio = 13.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Ethyl 1-[(6-chloropyridin-3-yl)methyl]-
3-phenyl-1H-pyrazole-5-carboxylate

In the title compound, $\text{C}_{18}\text{H}_{16}\text{ClN}_3\text{O}_2$, the dihedral angles made by the pyrazole ring with the pyridine and phenyl rings are $82.43(2)$ and $8.16(3)^\circ$, respectively. The crystal packing is stabilized mainly by van der Waals forces.

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Comment

The pyrazole unit is one of the core structures in a number of natural products. Many pyrazole derivatives are known to exhibit a wide range of biological properties such as anti-hyperglycaemic, analgesic, anti-inflammatory, antipyretic, antibacterial, hypoglycaemic, sedative-hypnotic (Cottineau *et al.*, 2002; Lee *et al.*, 2003), anticoagulant (Jia *et al.*, 2004) and antitumour (Wei *et al.*, 2006) activities. We report here the crystal structure of the title compound, (I).

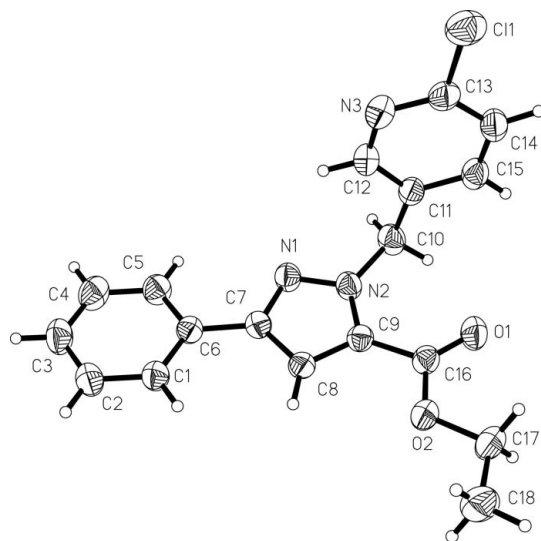


All bond lengths and angles in the molecule of (I) are normal (Allen *et al.*, 1987). The $\text{C}_{10}-\text{C}_{15}/\text{N}_3/\text{Cl}_1$ and C_1-C_6 planes make angles of $82.43(2)$ and $8.16(3)^\circ$, respectively, with the $\text{C}_6-\text{C}_9/\text{C}_{16}/\text{C}_{17}/\text{N}_1/\text{N}_2/\text{O}_1/\text{O}_2$ plane.

The crystal packing of (I) is stabilized mainly by van der Waals forces.

Experimental

A mixture of ethyl 3-phenyl-1H-pyrazole-5-carboxylate (0.01 mol), 2-chloro-5-(chloromethyl)pyridine (0.01 mol) and potassium carbonate (0.01 mol) in acetonitrile (20 ml) was heated to reflux for 4 h. The solvent was removed under reduced pressure and the residue extracted with ethyl acetate (30 ml). The organic phase was washed with brine and dried over anhydrous magnesium sulfate. After evaporation of the solvent under pressure, a solid was obtained (yield 82%). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a solution of the solid in ethyl acetate and petroleum ether (1:1 v/v) at room temperature for 3 d.

**Figure 1**

The molecular structure of compound (I), with displacement ellipsoids drawn at the 40% probability level.

Crystal data

$C_{18}H_{16}ClN_3O_2$
 $M_r = 341.79$
 Triclinic, $P\bar{1}$
 $a = 7.9796$ (9) Å
 $b = 9.9882$ (11) Å
 $c = 11.0724$ (13) Å
 $\alpha = 78.696$ (2)°
 $\beta = 88.823$ (2)°
 $\gamma = 77.216$ (2)°

$V = 843.68$ (17) Å³
 $Z = 2$
 $D_x = 1.345$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.24$ mm⁻¹
 $T = 298$ (2) K
 Block, colourless
 $0.56 \times 0.50 \times 0.20$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.877$, $T_{\max} = 0.953$

4207 measured reflections
 2947 independent reflections
 2590 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.013$
 $\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.115$
 $S = 1.07$
 2947 reflections
 217 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0576P)^2 + 0.1876P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.17$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³

All H atoms were placed in calculated positions, with C–H = 0.93–0.97 Å, and included in the final cycles of refinement using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aryl and methylene H atoms or $1.5U_{\text{eq}}(\text{C})$ for methyl H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Bruker, 1999); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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